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(54) DIELECTRIC PROCELAIN COMPOSITION AND ITS MANUFACTURE

(57) A dielectric porcelain composition. The conventional MgTiO_3 porcelain composition is not suitable material for small-sized electronic parts due to its relatively small dielectric constant (ϵ_r) of 20, while the CaTiO_3 porcelain composition is not suitable for electronic parts for high-frequency applications due to its small Q-value. In addition, it is difficult to control these conventional porcelain compositions' temperature coefficients of resonance frequency. This invention solves the above-mentioned problems by providing a dielectric porcelain composition expressed by the following chemical formula: $x\text{MgTiO}_3 \cdot (1-x)\text{CaTiO}_3 \cdot y(\text{Ln}^1_{1-w}\text{Ln}^2_w)_2\text{Ti}_{2z}\text{O}_{3+4z}$, where, Ln^1 and Ln^2 represent lanthanoid elements and $0 \leq w < 1$, $0.20 \leq x \leq 0.80$, $0.05 \leq y \leq 5.0$, and $0.25 \leq z \leq 1.5$. This dielectric porcelain composition can be used for resonators, filters, capacitors, etc., used for processing microwave-band signals.

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Description

FIELD OF THE INVENTION

5 The present invention relates to a ceramic dielectrics and a method for forming the same.

DESCRIPTION OF THE PRIOR ART

10 Recently, ceramic dielectrics for high frequency signals have been widely utilized as a material for a resonator used in an antenna duplexer of radio communication equipments such as a mobile phone, a cellular phone and a cordless telephone, a voltage-controlled oscillator (VCO) etc., or a filter used in a tuner for CATV. By using a material having a high dielectric constant for these, the wave length of an electromagnetic wave can be shortened to the length of $\epsilon_r^{-1/2}$ (ϵ_r : relative dielectric constant) under vacuum, so that one wave length, half a wave length or a quarter of a wave length, a resonant condition of an electromagnetic wave, can be shortened. Therefore, when the material is used, electronic
15 devices which process electric signals using a resonance of an electromagnetic wave can be easily downsized.

Characteristics required for the ceramic dielectrics for high frequency signals are:

- (1) Since the wave length of an electromagnetic wave is shortened to $\epsilon_r^{-1/2}$ in dielectrics, among ceramic dielectrics having the same resonant frequency, those which have a higher dielectric constant can be more downsized. Thus,
20 the dielectric constant should be as high as possible;
- (2) The dielectric loss ($1/Q$) should be small in the high-frequency-bandwidth, that is, the value of Q should be large; and
- (3) The variation of a resonant frequency depending on a temperature variation should be small, that is, the dependency on temperature of the relative dielectric constant (ϵ_r) should be small.

25 Besides, a reference clock frequency of electronic devices has been chosen in the microwave-bandwidth in many cases, and at present, the clock frequency of electronic equipments for the consumer is 1 GHz or so. However, as an increase of information content delivered per unit time, and a controlling speed of electronic equipments and a processing speed of signals becoming faster, electronic devices for a microwave which can be also used in a higher-frequency-bandwidth (several GHz) are becoming necessary. Accordingly, as a ceramic dielectrics constituting the electronic
30 devices for a microwave, a material having a large value of Q is required to make a microwave loss as small as possible. In addition, a material having a high dielectric constant is required to make the electronic devices as small as possible.

Hitherto, as ceramic dielectrics having the above characteristics, a MgTiO_3 -base and a CaTiO_3 -base ceramic dielectrics have been known.

35 The conventional MgTiO_3 -base ceramic dielectrics has a large value of Q , but has a relative dielectric constant (ϵ_r) of less than 20, which is not sufficiently high. As a result, it is difficult to downsize an element with the MgTiO_3 -base ceramic dielectrics therein. On the other hand, the CaTiO_3 -base ceramic dielectrics has a high dielectric constant, but has a small value of Q . As a result, it is not suitable for electronic devices processing signals in the high-frequency-bandwidth, and it is not suitable for being put to practical use because of its extremely high temperature coefficient of
40 resonant frequency (τ_f).

Moreover, since it is difficult to control the temperature coefficient of resonant frequency (τ_f) without changing the composition of these materials, it is required to change the temperature coefficient of resonant frequency (τ_f) by adding several kinds of additives such as rare earth elements to the above materials. But a ceramic dielectrics available for several requirements cannot be easily provided since the characteristics in changing compositions have not been systematically examined.
45

DISCLOSURE OF THE INVENTION

50 The present invention was achieved in order to solve these problems, and it is an object to provide a ceramic dielectrics having a larger value of Q and a higher relative dielectric constant (ϵ_r) than ever, whose temperature coefficient of resonant frequency (τ_f) can be controlled to an optional value in the range of +100 to -100 ppm/°C, and a method for forming the same.

A ceramic dielectrics (1) according to the present invention has a composition represented by $x\text{MgTiO}_3 \cdot (1-x)\text{CaTiO}_3 \cdot y(\text{Ln}^{1-w}\text{Ln}^{2-w})_2\text{Ti}_{2z}\text{O}_{3+4z}$ (wherein Ln^1 and Ln^2 are lanthanoids, and w , x , y , and z are values in the ranges
55 of $0 \leq w < 1$, $0.20 \leq x \leq 0.80$, $0.05 \leq y \leq 5.0$ and $0.25 \leq z \leq 1.5$, respectively.).

A ceramic dielectrics (2) according to the present invention substantially comprises a composition as a principal component represented by $x\text{MgTiO}_3 \cdot (1-x)\text{CaTiO}_3 \cdot y(\text{Ln}^{1-w}\text{Ln}^{2-w})_2\text{Ti}_{2z}\text{O}_{3+4z}$ (wherein Ln^1 and Ln^2 are lanthanoids, and w , x , y , and z are values in the ranges of $0 \leq w < 1$, $0.20 \leq x \leq 0.80$, $0.05 \leq y \leq 5.0$ and $0.25 \leq z \leq 1.5$, respectively.) and ZnO and/or MnO as additives in the range of a mol (where, $0 < a \leq 0.2$) to 1 mol of the principal component.

A ceramic dielectrics (3) according to the present invention has a composition represented by $x\text{MgTiO}_3 \cdot (1-x)\text{CaTiO}_3 \cdot y(\text{Nd}_{1-w}\text{Ln}^2_w)_2\text{Ti}_{2z}\text{O}_{3+4z}$ (wherein Ln^2 is a lanthanoid, and $w, x, y,$ and z are values in the ranges of $0 \leq w < 1, 0.20 \leq x \leq 0.80, 0.05 \leq y \leq 5.0$ and $0.25 \leq z \leq 1.5$, respectively).

A ceramic dielectrics (4) according to the present invention substantially comprises a composition as a principal component represented by $x\text{MgTiO}_3 \cdot (1-x)\text{CaTiO}_3 \cdot y(\text{Nd}_{1-w}\text{Ln}^2_w)_2\text{Ti}_{2z}\text{O}_{3+4z}$ (wherein Ln^2 is a lanthanoid, and $w, x, y,$ and z are values in the ranges of $0 \leq w < 1, 0.20 \leq x \leq 0.80, 0.05 \leq y \leq 5.0$ and $0.25 \leq z \leq 1.5$, respectively.) and ZnO and/or MnO as additives, in the range of a mol (where, $0 < a \leq 0.2$) to 1 mol of the principal component.

The above ceramic dielectrics (1)-(4) have a high relative dielectric constant (ϵ_r), 30-70, and a small dielectric loss because of a high value of Q, 3000 or more at a measuring frequency of 3 GHz, and by changing their compositions and so on, their temperature coefficient of resonant frequency (τ_f) can be controlled to a particular value in the range of +100 to -100 ppm / °C.

Besides, by making use of the electric characteristics of these ceramic dielectrics (1)-(4), a resonator for high frequency signals, a filter etc. can be sharply downsized.

A method for forming a ceramic dielectrics (1) according to the present invention comprises the steps of preparing raw materials selected from compounds each of which contains one of Mg, Ca, Ti, Ln^1 , and Ln^2 (here, Ln^1 and Ln^2 are lanthanoids) in such a proportion as a ceramic composition represented by $x\text{MgTiO}_3 \cdot (1-x)\text{CaTiO}_3 \cdot y(\text{Ln}^1_{1-w}\text{Ln}^2_w)_2\text{Ti}_{2z}\text{O}_{3+4z}$ (wherein Ln^1 and Ln^2 are lanthanoids, and $w, x, y,$ and z are values in the ranges of $0 \leq w < 1, 0.20 \leq x \leq 0.80, 0.05 \leq y \leq 5.0$ and $0.25 \leq z \leq 1.5$, respectively.) is formed after being sintered, mixing, pressing, and then sintering the same at a temperature of 1200-1600 °C in the air or an oxygen atmosphere.

A method for forming a ceramic dielectrics (2) according to the present invention comprises the steps of preparing raw materials selected from compounds each of which contains one of Mg, Ca, Ti, Ln^1 , and Ln^2 (here, Ln^1 and Ln^2 are lanthanoids) and a powder as sintering agents selected from compounds containing Zn and/or Mn in such a proportion as a ceramic composition substantially comprising a composition as a principal component represented by $x\text{MgTiO}_3 \cdot (1-x)\text{CaTiO}_3 \cdot y(\text{Ln}^1_{1-w}\text{Ln}^2_w)_2\text{Ti}_{2z}\text{O}_{3+4z}$ (wherein Ln^1 and Ln^2 are lanthanoids, and $w, x, y,$ and z are values in the ranges of $0 \leq w < 1, 0.20 \leq x \leq 0.80, 0.05 \leq y \leq 5.0$ and $0.25 \leq z \leq 1.5$, respectively.) and ZnO and/or MnO as additives in the range of a mol (where, $0 < a \leq 0.2$) to 1 mol of the principal component is formed after being sintered, mixing, pressing, and then sintering the same at a temperature of 1200-1600 °C in the air or an oxygen atmosphere.

A method for forming a ceramic dielectrics (3) according to the present invention comprises the steps of preparing raw materials selected from compounds each of which contains one of Mg, Ca, Ti, Ln^1 , and Ln^2 (here, Ln^1 and Ln^2 are lanthanoids) in such a proportion as a ceramic composition represented by $x\text{MgTiO}_3 \cdot (1-x)\text{CaTiO}_3 \cdot y(\text{Ln}^1_{1-w}\text{Ln}^2_w)_2\text{Ti}_{2z}\text{O}_{3+4z}$ (wherein Ln^1 and Ln^2 are lanthanoids, and $w, x, y,$ and z are values in the ranges of $0 \leq w < 1, 0.20 \leq x \leq 0.80, 0.05 \leq y \leq 5.0$ and $0.25 \leq z \leq 1.5$, respectively.) is formed after being sintered, mixing, calcining, granulating, pressing, and then sintering the same at a temperature of 1200-1600 °C in the air or an oxygen atmosphere.

A method for forming a ceramic dielectrics (4) according to the present invention comprises the steps of preparing raw materials selected from compounds each of which contains one of Mg, Ca, Ti, Ln^1 , and Ln^2 (here, Ln^1 and Ln^2 are lanthanoids) and a powder as sintering agents selected from compounds containing Zn and/or Mn in such a proportion as a ceramic composition substantially comprising a composition as a principal component represented by $x\text{MgTiO}_3 \cdot (1-x)\text{CaTiO}_3 \cdot y(\text{Ln}^1_{1-w}\text{Ln}^2_w)_2\text{Ti}_{2z}\text{O}_{3+4z}$ (wherein Ln^1 and Ln^2 are lanthanoids, and $w, x, y,$ and z are values in the ranges of $0 \leq w < 1, 0.20 \leq x \leq 0.80, 0.05 \leq y \leq 5.0$ and $0.25 \leq z \leq 1.5$, respectively.) and ZnO and/or MnO as additives in the range of a mol (where, $0 < a \leq 0.2$) to 1 mol of the principal component is formed after being sintered, mixing, calcining, granulating, pressing, and then sintering the same at a temperature of 1200-1600 °C in the air or an oxygen atmosphere.

A method for forming a ceramic dielectrics (5) according to the present invention comprises the steps of preparing raw materials selected from compounds each of which contains one of Mg, Ca, Ti, Nd, and Ln^2 (here, Ln^2 is a lanthanoid) in such a proportion as a ceramic composition represented by $x\text{MgTiO}_3 \cdot (1-x)\text{CaTiO}_3 \cdot y(\text{Nd}_{1-w}\text{Ln}^2_w)_2\text{Ti}_{2z}\text{O}_{3+4z}$ (wherein Ln^2 is a lanthanoid, and $w, x, y,$ and z are values in the ranges of $0 \leq w < 1, 0.20 \leq x \leq 0.80, 0.05 \leq y \leq 5.0$ and $0.25 \leq z \leq 1.5$, respectively.) is formed after being sintered, mixing, calcining, granulating, pressing, and then sintering the same at a temperature of 1200-1600 °C in the air or an oxygen atmosphere.

A method for forming a ceramic dielectrics (6) according to the present invention comprises the steps of preparing raw materials selected from compounds each of which contains one of Mg, Ca, Ti, Nd, and Ln^2 (here, Ln^2 is a lanthanoid) and a powder as sintering agents selected from compounds containing Zn and/or Mn in such a proportion as a ceramic composition substantially comprising a composition as a principal component represented by $x\text{MgTiO}_3 \cdot (1-x)\text{CaTiO}_3 \cdot y(\text{Nd}_{1-w}\text{Ln}^2_w)_2\text{Ti}_{2z}\text{O}_{3+4z}$ (wherein Ln^2 is a lanthanoid, and $w, x, y,$ and z are values in the ranges of $0 \leq w < 1, 0.20 \leq x \leq 0.80, 0.05 \leq y \leq 5.0$ and $0.25 \leq z \leq 1.5$, respectively.) and ZnO and/or MnO as additives in the range of a mol (where, $0 < a \leq 0.2$) to 1 mol of the principal component is formed after being sintered, mixing, calcining, granulating, pressing, and then sintering the same at a temperature of 1200-1600 °C in the air or an oxygen atmosphere.

According to the above methods for forming a ceramic dielectrics (1)-(6), a sintered body can have a uniform grain size and a ceramic dielectrics having a high relative dielectric constant (ϵ_r), 30-70, and a small dielectric loss because of a high value of Q, 3000 or more at a measuring frequency of 3 GHz, whose temperature coefficient of resonant fre-

quency (τ_r) can be controlled to a particular value in the range of +100 to -100 ppm/°C by changing their compositions etc., can be easily manufactured.

BRIEF DESCRIPTION OF THE DRAWINGS

Fig. 1(a) is a diagrammatic plain figure showing an apparatus used for measuring electric characteristics of the ceramic dielectrics according to Examples in the present invention, and Fig. 1(b) is a front view thereof.

FORM FOR CONDUCTING THE INVENTION

In the above methods for forming a ceramic dielectrics, raw materials are selected from compounds each of which contains one or two of Mg, Ca, Ti, and lanthanoids, and are prepared. In the case where addition of sintering agents is required, a powder selected from compounds containing Zn and Mn is added to the prepared raw materials. The compounds constituting these raw materials are not limited to oxides of the above elements and other compounds are available. For example, raw materials of carbonate, oxalate, nitrate, and alkoxide, which give oxides after being sintered, are exemplified. When oxides or carbonates of the above elements are used as raw materials, the average grain size of these compounds is preferably several μm or so. These raw materials are wet mixed in the usual way, and then dried, calcined, crushed, granulated, pressed and so on, so as to form a pressed body having a prescribed shape. After sintering the same, a ceramic dielectrics is obtained.

To be concrete, every raw material is accurately weighed in the above composition, which is wet mixed with balls, a well-known dispersant and distributed water in a pot mill for 24 hours or so, and finally a raw material mixture in a slurry is obtained. Next, the mixture in a slurry is dried and crushed. Then, the crushed powder may be transferred to a zirconia crucible for sintering, be calcined and synthesized at a temperature of 1000-1200 °C, and be crushed again, depending on necessity. As for the temperature condition of calcination and synthesis, the above range of temperature is preferable, since at a temperature of less than 1000 °C, a large amount of raw material is left and uniform sintering is prohibited, while at a temperature of over 1200 °C, sintering starts, which makes pulverizing difficult, and in both cases the value of Q tends to decrease.

Then, the crushed powder is granulated to a powder having an almost uniform grain size. After an organic binder etc. is added to the granulated powder, a pressed body having a prescribed shape is formed. As another way, a granulation treatment may be conducted by a spray drier etc. on the crushed powder with an organic binder etc. added thereto and a pressed body may be formed from the obtained powder.

Then, the ceramic dielectrics is obtained by defatting the obtained pressed body at a temperature of 600 °C or so, placing the defatted pressed body on a plate for sintering made of MgO etc., and sintering the pressed body placed thereon at a temperature of 1200-1600 °C in the air or an oxygen atmosphere for 2-8 hours. When the sintering temperature is less than 1200 °C, the pressed body is not sufficiently made fine, its value of Q becomes small, and its relative dielectric constant (ϵ_r) does not become high, while when the sintering temperature is over 1600 °C, the ceramic dielectrics itself becomes soft and the shape of the pressed body before being sintered cannot be kept. As a result, the above range of temperature is preferable.

The ceramic dielectrics formed by the above method has a high relative dielectric constant (ϵ_r), 30-70, and a large value of Q, 3000 or more at a measuring frequency of 3 GHz. In addition, it is possible to control its temperature coefficient of resonant frequency (τ_r) to a particular value in the range of +100 to -100 ppm/°C by changing x, y etc. in the composition formula. And the tissue of the ceramic dielectrics has an almost uniform grain size and is extremely fine, having a sintering density of 96.0-100% or so to the theoretical density, and is excellent in mechanical characteristics, so that it is suitable for use for a resonator or the like.

In the ceramic dielectrics, Ln^1 and Ln^2 are rare earth lanthanoid elements, and as the elements, La, Ce, Nd, Gd, Sm, and Dy are exemplified. As mentioned before, the temperature coefficient of resonant frequency (τ_r) can be controlled by changing the mixing ratio of these lanthanoids (y).

X is an atom ratio of Mg to a total amount of Mg and Ca in the ceramic dielectrics. When x is less than 0.20, Q becomes small, 3000 or less (at 3 GHz), and the temperature coefficient of resonant frequency (τ_r) becomes larger than 100, while when x is over 0.80, the relative dielectric constant (ϵ_r) becomes small, less than 30.

Y is a mol ratio of $(\text{Ln}^1_{1-w}\text{Ln}^2_w)_2\text{Ti}_{2z}\text{O}_{3+4z}$ to $(x\text{MgTiO}_3+(1-x)\text{CaTiO}_3)$. When y is less than 0.05, the temperature coefficient of resonant frequency (τ_r) becomes too large, while when y is over 5.0, Q becomes small, 3000 or less.

Z is a mol ratio of Ti to lanthanoids. When z is less than 0.25, Q becomes small, 3000 or less, while when z is over 1.5, the temperature coefficient of resonant frequency (τ_r) becomes too large.

A is mols of sintering agents to 1 mol of the principal component. The sintering agents contribute to a rise in Q when it is added in the case where sintering is difficult because of a small amount of Ti. On the other hand, the temperature coefficient of resonant frequency (τ_r) becomes small by adding the sintering agents. When a is over 0.20, Q becomes small, 3000 or less.

DESCRIPTION OF PREFERRED EMBODIMENTS

Examples and Comparative Examples of a ceramic dielectric and a method for forming the same according to the present invention are described below.

5 First, a method for forming a ceramic dielectrics according to Examples is explained.

Raw materials selected from MgO , CaCO_3 , TiO_2 , Ln^1_2O_3 , and Ln^2_2O_3 (Ln^1 and Ln^2 are lanthanoids), having an average grain size of some μm , were prepared in the proportions shown in Tables 1-5. Here, w, x, y, 2z, and a in Tables 1-5 correspond to the letters representing the ceramic dielectrics (1) and (2). In addition, as sintering agents, ZnO and MnCO_3 having the same average grain size as them were used.

10 A method for forming a sintered body is as described in the "FORM FOR CONDUCTING THE INVENTION", and as a raw material for pressing, powder calcined at 1000°C , and then crushed and granulated by addition of an organic binder etc. were used.

As for the composition of the ceramic dielectrics obtained by the above sintering, by conducting an ICP emission spectrochemical analysis after the ceramic dielectrics was dissolved in an acid, it was confirmed that the composition of the sintered body is the same as the preparation composition of the raw materials. For further observation of the structure of the sintered body, an etching treatment was conducted on the sintered body after being broken, and the surface thereof was observed with a scanning electron microscope (SEM). As a result, it was proved that the ceramic dielectrics according to Examples has a fine structure formed from particles having an almost uniform grain size.

And in order to use for measuring electric characteristics, a sample for the measurement of electric characteristics 20 was formed by sufficiently washing the obtained sintered body in deionized water, and then polishing it into such a shape as it has parallel main surfaces and a resonant frequency of 3 GHz.

Next, a method for measuring electric characteristics of the ceramic dielectrics according to Examples is explained.

As the electric characteristics, the resonant frequency, relative dielectric constant (ϵ_r) and Q were measured by the Hakki-Coleman dielectric resonator method.

25 Fig. 1(a) is a plain figure diagrammatically showing an apparatus used for measuring the electric characteristics, and Fig. 1(b) is a front view thereof.

A sample (ceramic dielectrics) 11, a subject of the measurement, is fixed in the state of being interposed between two parallel metal plates 12. 13 represents a column for keeping the metal plate stable on the sample.

In measuring the dielectric constant, the frequency characteristic was measured by oscillating a high frequency 30 from a probe 14 of a network analyzer, and the relative dielectric constant (ϵ_r) was calculated from the obtained resonant frequency peak in TE01 δ mode and the dimensions of a sample 11. As for measuring Q, the surface specific resistance of a metal plate 12 was found by using a standard sample, from which the dielectric loss for a metal plate 12 was found, and Q of a sample 11 was calculated by subtracting the dielectric loss for a metal plate 12 from the total dielectric loss. And the temperature coefficient of resonant frequency (τ_f) was measured in the atmospheric temperature range from -30 to $+85^\circ\text{C}$.

35 As the samples for the measurement, fifty samples of every Example (composition) were manufactured and the electric characteristics of each sample 11 were measured, and average values thereof were calculated. The results are shown in Tables 1-5.

Here, as Comparative Examples, the electric characteristics of the ceramic dielectrics having a composition outside 40 the range of the present invention, manufactured under the same conditions as Examples, and those of the ceramic dielectrics having a composition within the range of the present invention, manufactured at a sintering temperature of less than 1200°C or more than 1600°C ; were measured. The results are also shown in Tables 1-5. Comparative Examples are marked *.

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Table 1

Sample No	Composition of Ceramic Dielectrics						Sintering Temperature (°C)	Relative Dielectric Constant (ϵ_r)	Value of Q (at 3 GHz)	Temperature Coefficient τ_r (ppm/°C)	Note
	x	$L n^1$ (w=0)	y	2 z	Sintering Agents	a					
1	0.900	Nd	0.150	1.000		0.000	1400	20	50000	-15	*
2	0.800	Nd	0.150	1.000		0.000	1400	32	11000	9	
3	0.700	Nd	0.150	1.000		0.000	1400	42	8000	18	
4	0.600	Nd	0.150	1.000		0.000	1400	56	7300	31	
5	0.500	Nd	0.150	1.000		0.000	1400	62	6500	48	
6	0.400	Nd	0.150	1.000		0.000	1400	67	6200	56	
7	0.300	Nd	0.150	1.000		0.000	1400	69	6000	71	
8	0.200	Nd	0.150	1.000		0.000	1400	70	4700	85	
9	0.100	Nd	0.150	1.000		0.000	1400	71	1800	101	*
10	0.500	Nd	0.000	1.000		0.000	1400	72	1500	125	*
11	0.500	Nd	0.005	1.000		0.000	1400	69	3200	117	*
12	0.500	Nd	0.040	1.000		0.000	1400	68	1100	111	*
13	0.500	Nd	0.050	1.000		0.000	1400	67	3800	63	
14	0.500	Nd	0.250	1.000		0.000	1400	58	6300	40	
15	0.500	Nd	0.500	1.000		0.000	1400	52	6000	37	
16	0.500	Nd	1.000	1.000		0.000	1400	49	6700	19	
17	0.500	Nd	1.500	1.000		0.000	1400	46	6600	6	
18	0.500	Nd	2.500	1.000		0.000	1400	39	3500	-10	
19	0.500	Nd	5.000	1.000		0.000	1400	45	6000	-15	
20	0.500	Nd	5.300	1.000		0.000	1400	35	1000	—	*
21	0.500	Sm	0.005	1.000		0.000	1400	63	5700	109	*
22	0.500	Sm	0.040	1.000		0.000	1400	56	2300	105	*
23	0.500	Sm	0.050	1.000		0.000	1400	58	5200	62	
24	0.500	Sm	0.500	1.000		0.000	1400	50	5900	38	
25	0.500	Sm	1.000	1.000		0.000	1400	47	6200	18	
26	0.500	Sm	1.500	1.000		0.000	1400	42	5900	-15	
27	0.500	Sm	2.500	1.000		0.000	1400	31	3800	-24	
28	0.500	Sm	5.000	1.000		0.000	1400	30	3500	-15	
29	0.500	Sm	5.300	1.000		0.000	1400	24	1500	—	*

Table 2

Sample No.	Composition of Ceramic Dielectrics						Sintering Temperature (°C)	Relative Dielectric Constant (ϵ_r)	Value of Q (at 3 GHz)	Temperature Coefficient τ_r (ppm/°C)	Note
	x	L n ¹ (w=0)	y	2 z	Sintering Agents	a					
30	0.500	Gd	0.005	1.000		0.000	1400	61	1800	105	*
31	0.500	Gd	0.040	1.000		0.000	1400	60	1500	135	*
32	0.500	Gd	0.050	1.000		0.000	1400	56	4300	60	
33	0.500	Gd	0.500	1.000		0.000	1400	49	4900	37	
34	0.500	Gd	1.000	1.000		0.000	1400	45	5700	10	
35	0.500	Gd	1.500	1.000		0.000	1400	41	7200	21	
36	0.500	Gd	2.500	1.000		0.000	1400	31	3200	35	
37	0.500	Gd	5.000	1.000		0.000	1400	30	3700	-35	
38	0.500	Gd	5.300	1.000		0.000	1400	20	1300	40	*
39	0.500	Dy	0.005	1.000		0.000	1400	60	1300	104	*
40	0.500	Dy	0.040	1.000		0.000	1400	57	500	120	*
41	0.500	Dy	0.050	1.000		0.000	1400	55	4900	58	
42	0.500	Dy	0.500	1.000		0.000	1400	48	5500	36	
43	0.500	Dy	1.000	1.000		0.000	1400	44	6200	10	
44	0.500	Dy	1.500	1.000		0.000	1400	40	6100	-23	
45	0.500	Dy	2.500	1.000		0.000	1400	34	3200	-61	
46	0.500	Dy	5.000	1.000		0.000	1400	30	3200	-47	
47	0.500	Dy	5.300	1.000		0.000	1400	27	1800	—	*
48	0.500	Ce	0.005	1.000		0.000	1400	70	1300	121	*
49	0.500	Ce	0.050	1.000		0.000	1400	68	3900	65	
50	0.500	Ce	0.500	1.000		0.000	1400	59	4300	42	
51	0.500	Ce	1.000	1.000		0.000	1400	53	5100	21	
52	0.500	Ce	1.500	1.000		0.000	1400	45	4200	4	
53	0.500	Ce	2.500	1.000		0.000	1400	33	4000	-15	
54	0.500	Nd	0.150	1.000	MnO	0.010	1400	61	6600	49	
55	0.500	Nd	0.150	1.000	MnO	0.100	1400	60	6500	47	
56	0.500	Nd	0.150	1.000	MnO	0.180	1400	55	6800	40	
57	0.500	Nd	0.150	1.000	MnO	0.210	1400	—	—	—	*
58	0.500	Nd	0.150	1.000	ZnO	0.010	1400	60	6900	47	
59	0.500	Nd	0.150	1.000	ZnO	0.100	1400	58	7000	45	
60	0.500	Nd	0.150	1.000	ZnO	0.180	1400	53	7200	39	
61	0.500	Nd	0.150	1.000	ZnO	0.210	1400	—	800以下	—	*

Table 3

Sample No.	Composition of Ceramic Dielectrics						Sintering Temperature (°C)	Relative Dielectric Constant (ϵ_r)	Value of Q (at 3 GHz)	Temperature Coefficient τ_r (ppm/°C)	Note
	x	L n ¹ (w=0)	y	2 z	Sintering Agents	a					
62	0.500	Nd	0.150	0.400		0.000	1400	28	1600	35	*
63	0.500	Nd	0.150	0.500		0.000	1400	60	6400	21	
64	0.500	Nd	0.150	0.750		0.000	1400	61	6700	50	
65	0.500	Nd	0.150	1.250		0.000	1400	64	6500	40	
66	0.500	Nd	0.150	1.500		0.000	1400	62	5200	39	
67	0.500	Nd	0.150	2.000		0.000	1400	63	5400	37	
68	0.500	Nd	0.150	3.100		0.000	1400	76	4600	135	*
69	0.500	Nd	0.150	1.000		0.000	1100	23	800以下	—	*
70	0.500	Nd	0.150	1.000		0.000	1150	27	900	—	*
71	0.500	Nd	0.150	1.000		0.000	1200	60	6000	41	
72	0.500	Nd	0.150	1.000		0.000	1300	61	5700	47	
73	0.500	Nd	0.150	1.000		0.000	1500	62	6200	48	
74	0.500	Nd	0.150	1.000		0.000	1600	63	5600	49	
75	0.500	Nd	0.150	1.000		0.000	1650	unmeasured owing to dissolution			*
76	0.500	Nd	0.150	1.000	MnO	0.100	1150	29	1100	47	*
77	0.500	Nd	0.150	1.000	MnO	0.100	1200	60	6500	41	
78	0.500	Nd	0.150	1.000	MnO	0.100	1300	60	6000	41	
79	0.500	Nd	0.150	1.000	ZnO	0.100	1150	27	900	44	*
80	0.500	Nd	0.150	1.000	ZnO	0.100	1200	59	6800	46	
81	0.500	Nd	0.150	1.000	ZnO	0.100	1300	60	7100	47	
82	0.500	La	0.500	1.000		0.000	1400	70	4900	93	
83	0.500	La	0	1.000		0.000	1400	57	3400	800	*
84	0.500	La	0.001	1.000		0.000	1400	61	3900	148	*
85	0.500	La	0.010	1.000		0.000	1400	52	4200	125	*
86	0.500	La	0.100	1.000		0.000	1400	49	4300	69	
87	0.500	La	1.000	1.000		0.000	1400	51	4700	63	
88	0.500	La	3.000	1.000		0.000	1400	47	3500	21	
89	0.500	La	4.900	1.000		0.000	1400	43	3400	11	
90	0.500	La	5.500	1.000		0.000	1400	41	360	-100	*
91	0.700	La	0.500	1.000		0.000	1400	64	6300	93	
92	0.700	La	2.500	1.000		0.000	1400	59	3800	19	
93	0.700	La	6.000	1.000		0.000	1400	43	560	0	*
94	1.000	La	3.000	1.000		0.000	1400	85	490	490	*
95	1.000	La	6.000	1.000		0.000	1400	47	300	580	*

Table 4

Sample No.	Composition of Ceramic Dielectrics							Sintering Temperature (°C)	Relative Dielectric Constant (ϵ_r)	Value of Q (at 3 GHz)	Temperature Coefficient τ_r (ppm/°C)	Note
	x	Ln ¹ Ln ²	w	y	2 z	Sintering Agents	a					
96	0.500	Nd Sm	0.500	0.500	1.000		0.000	1400	63	4800	62	
97	0.500	Nd Gd	0.500	0.500	1.000		0.000	1400	60	4100	61	
98	0.500	Nd Dy	0.500	0.500	1.000		0.000	1400	60	4200	60	
99	0.500	Nd Ce	0.500	0.500	1.000		0.000	1400	67	3600	64	
100	0.500	Sm Gd	0.500	0.500	1.000		0.000	1400	58	4900	61	
101	0.500	Sm Dy	0.500	0.500	1.000		0.000	1400	60	4200	60	
102	0.500	Sm Ce	0.500	0.500	1.000		0.000	1400	57	4200	63	
103	0.500	Gd Dy	0.500	0.500	1.000		0.000	1400	51	4500	59	
104	0.500	Gd Ce	0.500	0.500	1.000		0.000	1400	60	4000	63	
105	0.500	Dy Ce	0.500	0.500	1.000		0.000	1400	61	4200	51	
106	0.500	Nd La	0.010	0.500	1.000		0.000	1400	52	10000	25	
107	0.500	Nd La	0.100	0.500	1.000		0.000	1400	56	11500	25	
108	0.500	Nd La	0.200	0.500	1.000		0.000	1400	59	12100	37	
109	0.500	Nd La	0.300	0.500	1.000		0.000	1400	61	11300	44	
110	0.500	Nd La	0.400	0.500	1.000		0.000	1400	64	10800	56	
111	0.500	Nd La	0.500	0.500	1.000		0.000	1400	66	9700	63	

Table 5

Sample No.	Composition of Ceramic Dielectrics							Sintering Temperature (°C)	Relative Dielectric Constant (ϵ_r)	Value of Q (at 3 GHz)	Temperature Coefficient τ_f (ppm/°C)	Note
	x	Ln^1 Ln^2	w	y	2 z	Sintering Agents	a					
112	0.500	Nd La	0.600	0.500	1.000		0.000	1400	66	8300	65	
113	0.500	Nd La	0.700	0.500	1.000		0.000	1400	68	6200	73	
114	0.500	Nd La	0.800	0.500	1.000		0.000	1400	69	5000	77	
115	0.500	Nd La	0.900	0.500	1.000		0.000	1400	70	4800	91	
116	0.500	Nd La	0.500	0	1.000		0.000	1400	56	3500	780	*
117	0.500	Nd La	0.500	0.001	1.000		0.000	1400	57	4100	190	*
118	0.500	Nd La	0.500	0.010	1.000		0.000	1400	55	5100	160	*
119	0.500	Nd La	0.500	0.100	1.000		0.000	1400	54	4100	71	
120	0.500	Nd La	0.500	1.000	1.000		0.000	1400	49	4300	53	
121	0.500	Nd La	0.500	3.000	1.000		0.000	1400	46	3900	18	
122	0.500	Nd La	0.500	4.500	1.000		0.000	1400	43	3300	-3	
123	0.500	Nd La	0.500	5.500	1.000		0.000	1400	35	870	-105	*
124	0.700	Nd La	0.500	0.500	1.000		0.000	1400	65	6900	78	
125	0.700	Nd La	0.500	4.500	1.000		0.000	1400	46	3300	11	
126	0.700	Nd La	0.500	6.000	1.000		0.000	1400	44	950	8	*

As obvious from the results in Tables 1-5, the ceramic dielectrics according to Examples have a high relative dielectric constant (ϵ_r), 30-70, and a small dielectric loss because of a large value of Q, 3000 or more at a measuring frequency of 3 GHz, and the temperature coefficient of resonant frequency (τ_f) can be controlled to be a particular value within the range of +100 to -100 ppm / °C by changing the ratio (x) of MgO to CaO, or the ratio (y) of $(x\text{MgTiO}_3 \cdot (1-x)\text{CaTiO}_3)$ to $(\text{Ln}^{1-1-w}\text{Ln}^2_w)_2\text{Ti}_{2z}\text{O}_{3+4z}$. And by setting the sintering temperature at a temperature of 1200-1600 °C, a ceramic dielectrics having the above excellent electric characteristics can be manufactured.

On the other hand, among the ceramic dielectrics according to Comparative Examples, those having x of less than 0.20 or more than 0.80, y of less than 0.05 or more than 5.0, z of less than 0.25 or more than 1.5, and a of more than 0.2, had at least one of Q, relative dielectric constant (ϵ_r) and temperature coefficient of resonant frequency (τ_f) outside the above ranges. As a result, it is difficult to use them as a material for a resonator, a filter or the like.

And as for the ceramic dielectrics formed at a sintering temperature of less than 1200 °C or more than 1600 °C, the obtained electric characteristics thereof were not desirable, either.

POSSIBILITY OF INDUSTRIAL APPLICATION

A ceramic dielectrics according to the present invention can be used as a material for a high-performance resonator processing a signal in the microwave-bandwidth, a filter, a capacitor or the like.

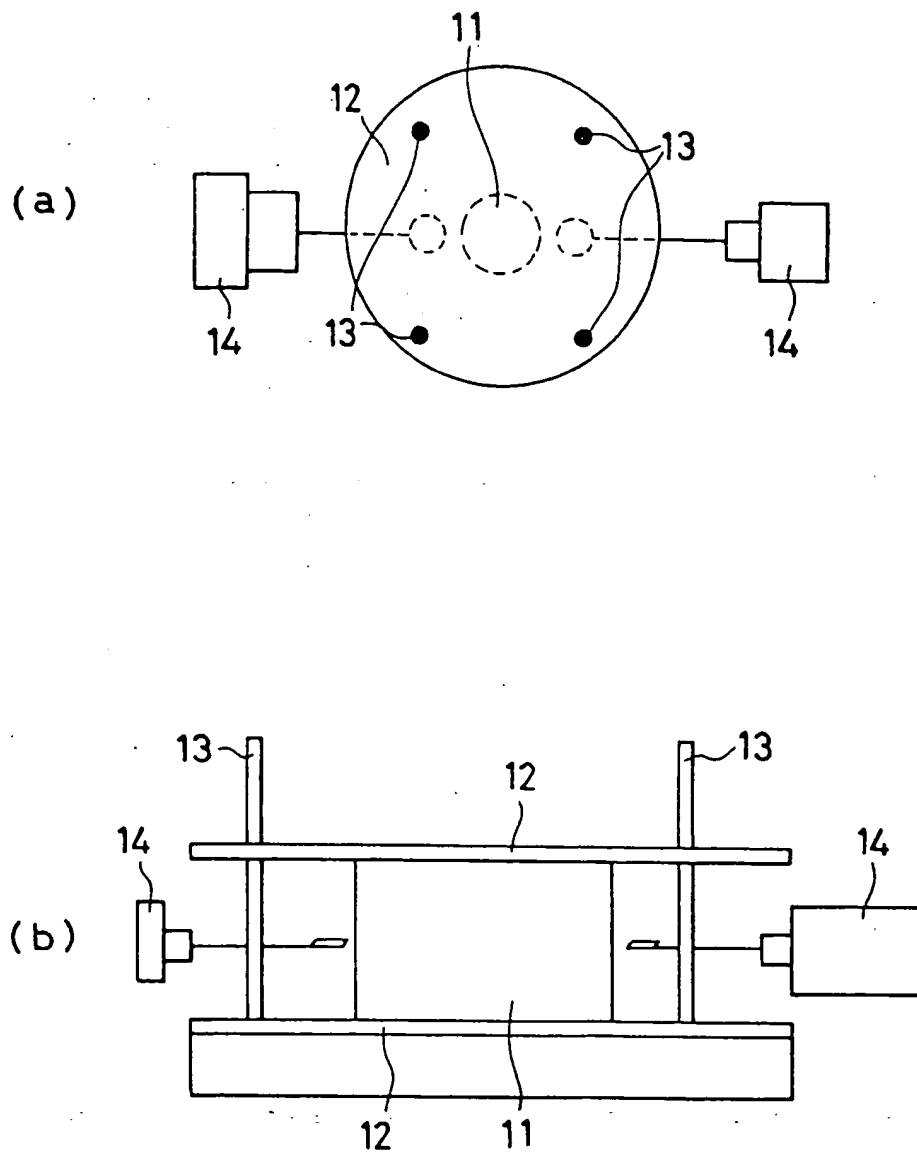
Claims

1. A ceramic dielectrics, having a composition represented by $x\text{MgTiO}_3 \cdot (1-x)\text{CaTiO}_3 \cdot y(\text{Ln}^1_{1-w}\text{Ln}^2_w)_2\text{Ti}_{2z}\text{O}_{3+4z}$ (wherein Ln^1 and Ln^2 are lanthanoids and $w, x, y,$ and z are values in the ranges of $0 \leq w < 1, 0.20 \leq x \leq 0.80, 0.05 \leq y \leq 5.0,$ and $0.25 \leq z \leq 1.5,$ respectively).
2. A ceramic dielectrics, substantially comprising a composition as a principal component represented by $x\text{MgTiO}_3 \cdot (1-x)\text{CaTiO}_3 \cdot y(\text{Ln}^1_{1-w}\text{Ln}^2_w)_2\text{Ti}_{2z}\text{O}_{3+4z}$ (wherein Ln^1 and Ln^2 are lanthanoids and $w, x, y,$ and z are values in the ranges of $0 \leq w < 1, 0.20 \leq x \leq 0.80, 0.05 \leq y \leq 5.0,$ and $0.25 \leq z \leq 1.5,$ respectively) and ZnO and/or MnO as additives in the range of a mol (where, $0 < a \leq 0.2$) to 1 mol of the principal component.
3. A ceramic dielectrics, having a composition represented by $x\text{MgTiO}_3 \cdot (1-x)\text{CaTiO}_3 \cdot y(\text{Nd}_{1-w}\text{Ln}^2_w)_2\text{Ti}_{2z}\text{O}_{3+4z}$ (wherein Ln^2 is a lanthanoid and $w, x, y,$ and z are values in the ranges of $0 \leq w < 1, 0.20 \leq x \leq 0.80, 0.05 \leq y \leq 5.0,$ and $0.25 \leq z \leq 1.5,$ respectively).
4. A ceramic dielectrics, substantially comprising a composition as a principal component represented by $x\text{MgTiO}_3 \cdot (1-x)\text{CaTiO}_3 \cdot y(\text{Nd}_{1-w}\text{Ln}^2_w)_2\text{Ti}_{2z}\text{O}_{3+4z}$ (wherein Ln^2 is a lanthanoid and $w, x, y,$ and z are values in the ranges of $0 \leq w < 1, 0.20 \leq x \leq 0.80, 0.05 \leq y \leq 5.0,$ and $0.25 \leq z \leq 1.5,$ respectively) and ZnO and/or MnO as additives in the range of a mol (where, $0 < a \leq 0.2$) to 1 mol of the principal component.
5. A method for forming a ceramic dielectrics, comprising the steps of preparing raw materials selected from compounds each of which contains one of Mg, Ca, Ti, $\text{Ln}^1,$ and Ln^2 (wherein Ln^1 and Ln^2 are lanthanoids) in such a proportion as a ceramic composition represented by $x\text{MgTiO}_3 \cdot (1-x)\text{CaTiO}_3 \cdot y(\text{Ln}^1_{1-w}\text{Ln}^2_w)_2\text{Ti}_{2z}\text{O}_{3+4z}$ (wherein Ln^1 and Ln^2 are lanthanoids and $w, x, y,$ and z are values in the ranges of $0 \leq w < 1, 0.20 \leq x \leq 0.80, 0.05 \leq y \leq 5.0,$ and $0.25 \leq z \leq 1.5,$ respectively) is formed after being sintered, mixing, pressing, and then sintering the same at a temperature of 1200-1600 °C in the air or an oxygen atmosphere.
6. A method for forming a ceramic dielectrics, comprising the steps of preparing raw materials selected from compounds each of which contains one of Mg, Ca, Ti, $\text{Ln}^1,$ and Ln^2 (wherein Ln^1 and Ln^2 are lanthanoids) and a powder as sintering agents selected from compounds containing Zn and/or Mn in such a proportion as a ceramic composition substantially comprising a composition as a principal component represented by $x\text{MgTiO}_3 \cdot (1-x)\text{CaTiO}_3 \cdot y(\text{Ln}^1_{1-w}\text{Ln}^2_w)_2\text{Ti}_{2z}\text{O}_{3+4z}$ (wherein Ln^1 and Ln^2 are lanthanoids and $w, x, y,$ and z are values in the ranges of $0 \leq w < 1, 0.20 \leq x \leq 0.80, 0.05 \leq y \leq 5.0,$ and $0.25 \leq z \leq 1.5,$ respectively) and ZnO and/or MnO as additives in the range of a mol (where, $0 < a \leq 0.2$) to 1 mol of the principal component is formed after being sintered, mixing, pressing, and then sintering the same at a temperature of 1200-1600 °C in the air or an oxygen atmosphere.
7. A method for forming a ceramic dielectrics, comprising the steps of preparing raw materials selected from compounds each of which contains one of Mg, Ca, Ti, $\text{Ln}^1,$ and Ln^2 (wherein Ln^1 and Ln^2 are lanthanoids) in such a proportion as a ceramic composition represented by $x\text{MgTiO}_3 \cdot (1-x)\text{CaTiO}_3 \cdot y(\text{Ln}^1_{1-w}\text{Ln}^2_w)_2\text{Ti}_{2z}\text{O}_{3+4z}$ (wherein Ln^1 and Ln^2 are lanthanoids and $w, x, y,$ and z are values in the ranges of $0 \leq w < 1, 0.20 \leq x \leq 0.80, 0.05 \leq y \leq 5.0,$ and $0.25 \leq z \leq 1.5,$ respectively) is formed after being sintered, mixing, calcining, granulating, pressing, and then sintering the same at a temperature of 1200-1600 °C in the air or an oxygen atmosphere.
8. A method for forming a ceramic dielectrics, comprising the steps of preparing raw materials selected from compounds each of which contains one of Mg, Ca, Ti, $\text{Ln}^1,$ and Ln^2 (wherein Ln^1 and Ln^2 are lanthanoids) and a powder as sintering agents selected from compounds containing Zn and/or Mn in such a proportion as a ceramic composition substantially comprising a composition as a principal component represented by $x\text{MgTiO}_3 \cdot (1-x)\text{CaTiO}_3 \cdot y(\text{Ln}^1_{1-w}\text{Ln}^2_w)_2\text{Ti}_{2z}\text{O}_{3+4z}$ (wherein Ln^1 and Ln^2 are lanthanoids and $w, x, y,$ and z are values in the ranges of $0 \leq w < 1, 0.20 \leq x \leq 0.80, 0.05 \leq y \leq 5.0,$ and $0.25 \leq z \leq 1.5,$ respectively) and ZnO and/or MnO as additives in the range of a mol (where, $0 < a \leq 0.2$) to 1 mol of the principal component is formed after being sin-

tered, mixing, calcining, granulating, pressing, and then sintering the same at a temperature of 1200-1600 °C in the air or an oxygen atmosphere.

- 5 9. A method for forming a ceramic dielectrics, comprising the steps of preparing raw materials selected from compounds each of which contains one of Mg, Ca, Ti, Nd, and Ln^2 (wherein Ln^2 is a lanthanoid) in such a proportion as a ceramic composition represented by $x\text{MgTiO}_3 \cdot (1-x)\text{CaTiO}_3 \cdot y(\text{Nd}_{1-w}\text{Ln}_w^2)_2\text{Ti}_{2z}\text{O}_{3+4z}$ (wherein Ln^2 is a lanthanoid and w, x, y, and z are values in the ranges of $0 \leq w < 1$, $0.20 \leq x \leq 0.80$, $0.05 \leq y \leq 5.0$, and $0.25 \leq z \leq 1.5$, respectively) is formed after being sintered, mixing, calcining, granulating, pressing, and then sintering the same at a temperature of 1200-1600 °C in the air or an oxygen atmosphere.
- 10 10. A method for forming a ceramic dielectrics, comprising the steps of preparing raw materials selected from compounds each of which contains one of Mg, Ca, Ti, Nd, and Ln^2 (wherein Ln^2 is a lanthanoid) and a powder as sintering agents selected from compounds containing Zn and/or Mn in such a proportion as a ceramic composition substantially comprising a composition as a principal component represented by $x\text{MgTiO}_3 \cdot (1-x)\text{CaTiO}_3 \cdot y(\text{Nd}_{1-w}\text{Ln}_w^2)_2\text{Ti}_{2z}\text{O}_{3+4z}$ (wherein Ln^2 is a lanthanoid and w, x, y, and z are values in the ranges of $0 \leq w < 1$, $0.20 \leq x \leq 0.80$, $0.05 \leq y \leq 5.0$, and $0.25 \leq z \leq 1.5$, respectively) and ZnO and/or MnO as additives in the range of a mol (where, $0 < a \leq 0.2$) to 1 mol of the principal component is formed after being sintered, mixing, calcining, granulating, pressing, and then sintering the same at a temperature of 1200-1600 °C in the air or an oxygen atmosphere.
- 15 20 25 30 35 40 45 50 55

Fig. 1



INTERNATIONAL SEARCH REPORT

International application No.

PCT/JP95/01784

A. CLASSIFICATION OF SUBJECT MATTER

Int. Cl⁶ H01B3/12

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

Int. Cl⁶ H01B3/12, C04B35/00

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Jitsuyo Shinan Koho 1926 - 1993
 Kokai Jitsuyo Shinan Koho 1971 - 1993

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X Y	JP, 56-54272, A (TDK Corp.), May 14, 1981 (14. 05. 81) (Family: none)	1, 2, 4, 5 4, 6-10
X Y	JP, 51-67999, A (TDK Corp.), June 12, 1976 (12. 06. 76) (Family: none)	1 - 5 4, 6-10
Y	JP, 63-151655, A (Murata Mfg. Co., Ltd.), December 16, 1986 (16. 12. 86) (Family: none)	4, 6-10

☐ Further documents are listed in the continuation of Box C.☐ See patent family annex.

* Special categories of cited documents:

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Date of the actual completion of the international search
 November 17, 1995 (17. 11. 95)

Date of mailing of the international search report
 December 5, 1995 (05. 12. 95)

Name and mailing address of the ISA/
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